


Name:			
Enrolment No:			
UNIVERSITY OF PETROLEUM AND ENERGY STUDIES End Semester Examination, December 2022			
Course: Engineering Thermodynamics Program: B Tech (Mechatronics Engineering) Course Code: MECH2014		Semester: III Time : 03 hrs. Max. Marks: 100	
Instructions: Use of Steam Table is allowed.			
SECTION A (5Qx4M=20Marks)			
S. No.		Marks	CO
Q 1	What do you mean by thermodynamic equilibrium?	4	CO1
Q 2	Discuss the conditions which must be fulfilled by a reversible process. Give some examples of ideal reversible processes.	4	CO1
Q 3	In a cyclic process, heat transfers are + 14.7 kJ, – 25.2 kJ, – 3.56 kJ and +31.5 kJ. What is the network for this cyclic process?	4	CO2
Q 4	Define entropy. What do you understand by entropy principle? What are the causes of entropy increases?	4	CO1
Q 5	Give the following statements of second law of thermodynamics. (i) Clausius statement (ii) Kelvin-Planck statement.	4	CO1
SECTION B (4Qx10M= 40 Marks)			
Q 6	A flow rate of 0.42 kg/s is maintained in a steam turbine under steady flow conditions as it receives steam with an enthalpy 3240 kJ/kg, velocity 35 m/s and elevation 4 m. The outlet of steam from the turbine has enthalpy of 2450 kJ/kg, velocity 125 m/s, and elevation 1 m. In the entire process the heat lost takes place at the rate of 0.25 kJ/s. Determine the power output of the turbine in kW?	10	CO2
Q 7	A mass of gas is compressed in a quasi-static process from 70 kPa, 0.1 m ³ to 0.4 MPa, 0.03 m ³ . Assuming that the pressure and volume are related by $PV^n = \text{constant}$, find the work done by the gas system.	10	CO2
Q 8	A refrigerator having a COP of 5 is driven by an engine having 30% thermal efficiency. What is the heat supplied to the engine for 1 MJ of heat removed from the cold body by the refrigerator? If this system is used as a heat pump, how many MJ of heat would be available for heating for each MJ of heat input to the engine?	10	CO2
Q 9	Draw and explain the Rankine cycle and explain the working principle with all the components.	10	CO3
OR			

	Discuss the working of a four stroke Diesel engine. With the help of pressure-volume diagram, explain the thermodynamic processes involved.		
SECTION-C (2Qx20M=40 Marks)			
Q 10	Derive an expression of thermal efficiency of an Otto cycle. List down the assumptions made in an Otto cycle.	20	CO3
Q 11	<p>A steam boiler generates steam at 30 bar, 300 °C at the rate of 2 kg/s. This steam is expanded isentropically in a turbine to a condenser pressure of 0.05 bar, condensed at constant pressure and pumped back to boiler.</p> <p>i) Find the heat supplied in the boiler per hour. ii) Determine the quality of steam after expansion. iii) What is the power generated by the turbine? Estimate the Rankine efficiency considering pump work.</p> <p style="text-align: center;">OR</p> <p>A steam turbine receives steam at pressure 20 bar and degree of superheat of 88.6°C. The exhaust pressure is 0.07 bar and the expansion of steam takes place isentropically Using steam table, calculate the following:</p> <p>i) Heat supplied (feed pump supplies water to the boiler at 20 bar) ii) Work done by the turbine.</p>	20	CO4